



# Szuperparamágneses vas-oxid nanorészecskék (SPIONs) teranosztikai célú fejlesztése

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Vizes Kolloidok Kutatócsoporthoz  
Fizikai Kémiai és Anyagtudományi Tanszék  
Szegedi Tudományegyetem



# Biomedical applications of magnetic nanoparticles

The most important applications:

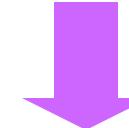
**MRI contrast agent** – **diagnosis**

**magnetic cell labeling and separation** - **therapy**

**magnetic hyperthermia** - **therapy**

**targeted drug delivery** - **nanomedicine**

**Combination of therapy  
and diagnosis**



**theranostic agents**

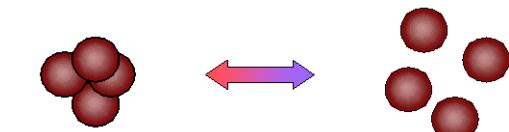
**Criteria:** these applications require the magnetic nanoparticles to be **non-toxic**,

**chemically stable**,

**uniform in size**, and

**well-dispersed in aqueous media.**

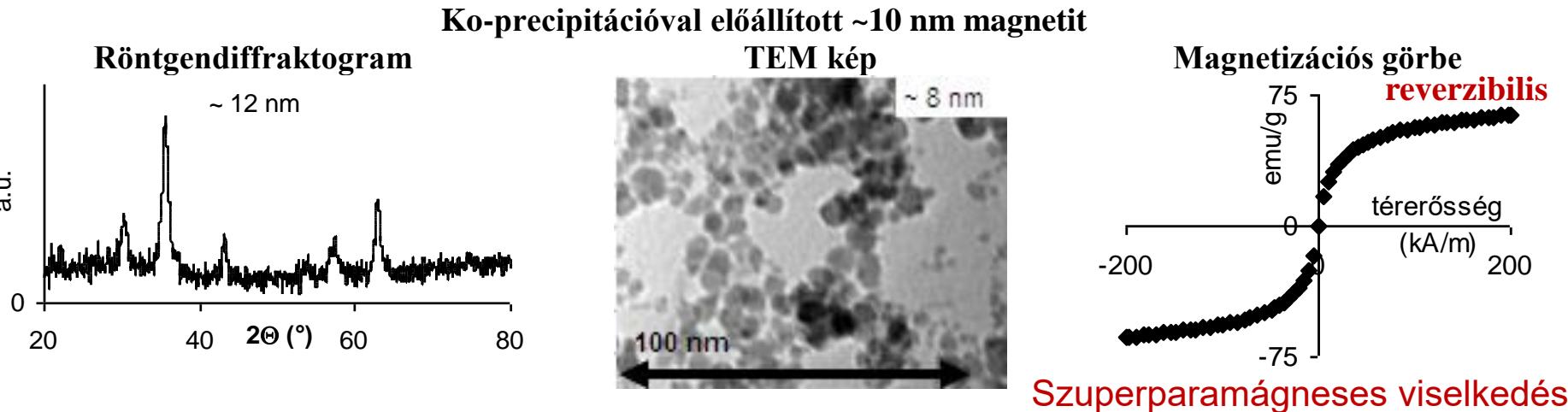
Magnetic nanoparticles are of magnetite and maghemite dominantly, because living systems know what to do with them, **iron oxides are excreted via the liver** after the treatment.



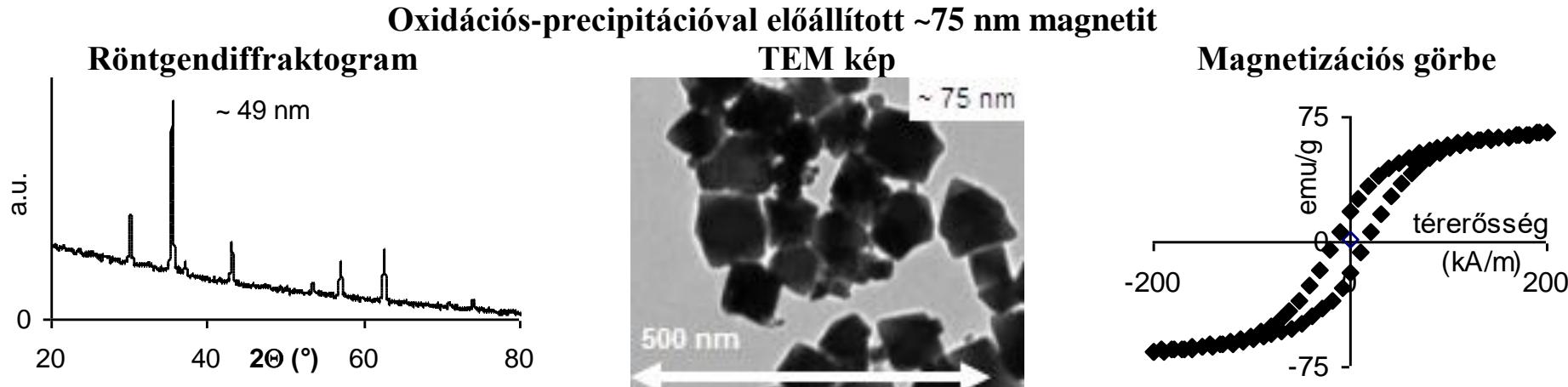
**Particle aggregation** must be excluded in magnetic field during application with reference to the **danger of embolism** in blood vessel.

# Mágneses vas-oxid részecskék szintetizálása: méretfüggés a nano mérettartományban

Ko-precipitáció ( $\text{Fe}^{2+} + 2\text{Fe}^{3+} + 4\text{OH}^- \Rightarrow \text{Fe}_3\text{O}_4 + 4\text{H}^+$ ) + hidrotermális öregítés)



Oxidációs-precipitáció ( $\text{Fe(OH)}_2$  zöld rozsda  $\Rightarrow \text{Fe}_3\text{O}_4$  magnetit)

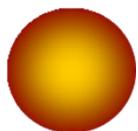


# Post-coating of SPIONs with polycarboxylates

- via thermodynamically driven spontaneous multiple binding to  $\equiv\text{Fe-OH}$  sites,
- providing combined electro-steric stabilization, hydrophilicity and
- free carboxylate sites for anchoring bioactive molecules via peptide bonding.

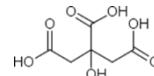
**naked SPIONs:**

preparation  
purification  
characterization

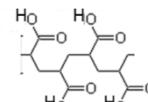


+

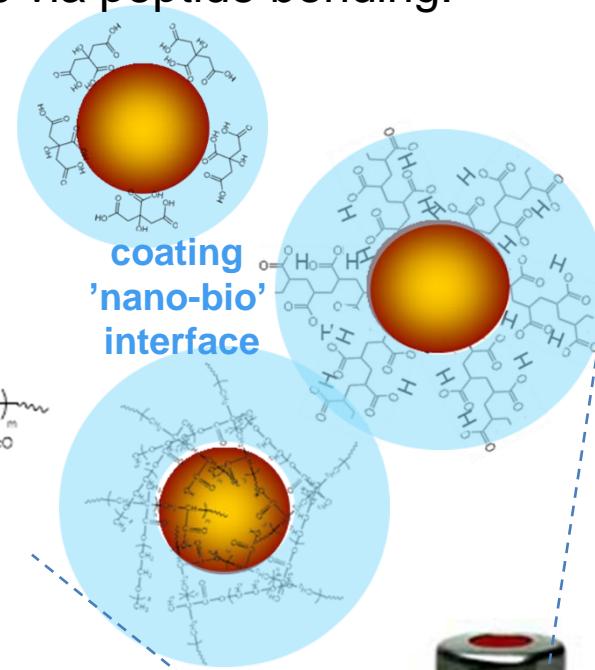
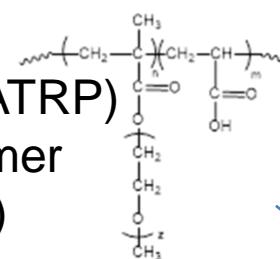
citric acid (CA)



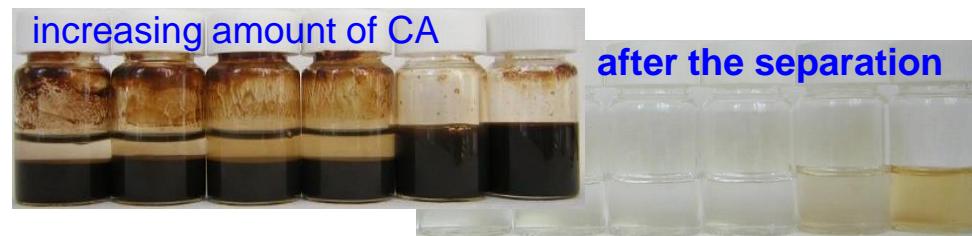
commercial  
poly(acrylic acid)s  
(PAA, PAM, etc.)



synthesized (via ATRP)  
comb-like copolymer  
P(PEGMA-co-AA)



# Characterization: chemical stability of carboxylated SPIONS



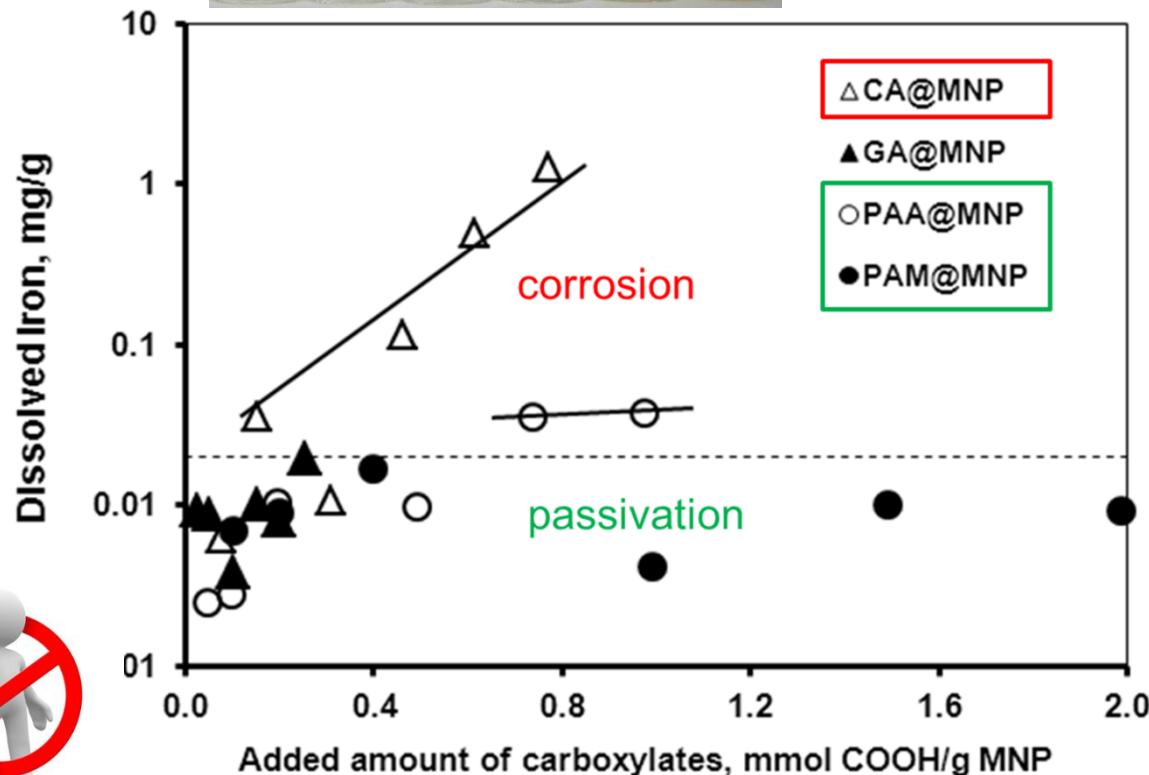
in the presence of citric acid (CA)

- iron oxide cores dissolve
- Fe(II)/Fe(III) ions leak into the medium

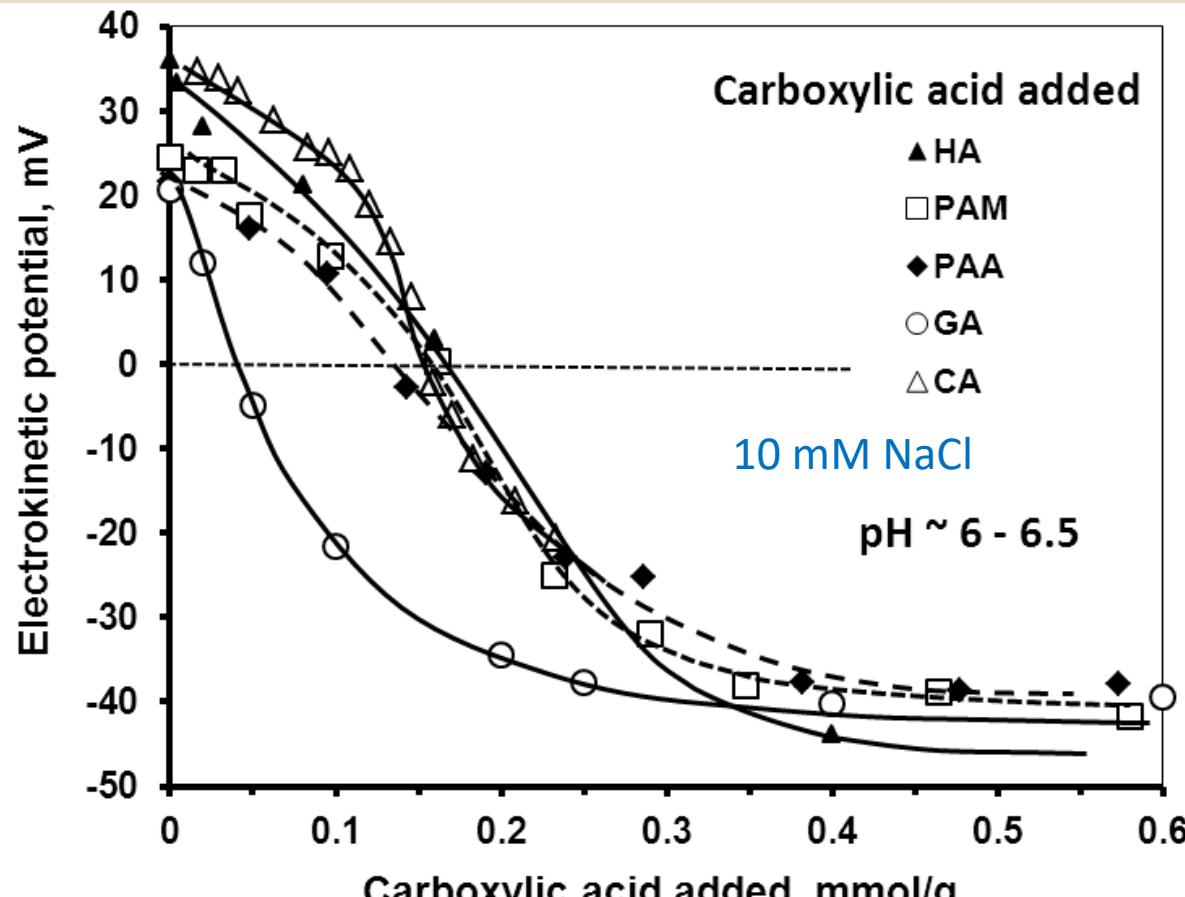
while polyelectrolyte coatings  
(PAA, PAM) prevent dissolution

Therefore **citrated SPIONS**  
(**CA@MNP**)

**should be ruled out from**  
**in vitro/vivo tests and**  
**biomedical application.**



# Characterization of particle charge: electrokinetic ( $\zeta$ ) potential



Magnetite

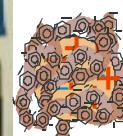


Increasing amount of HA  $\Rightarrow$  charge neutralization  $\Rightarrow$  charge reversal destabilization restabilization

solution conditions:  
pH, ionic strength,  
buffer (quality and  
concentration), etc.

Effect of organic acids

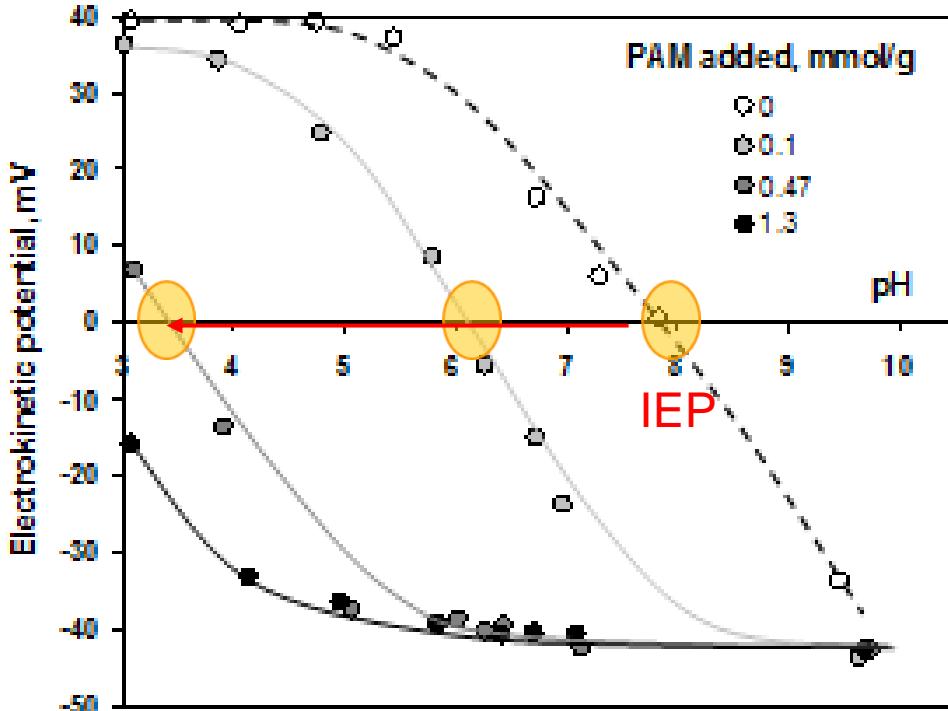
quantitative comparision for  
surface charge  
neutralization



overcharging, if the amount of  
organic polyacids exceeds the  
amount of oppositely charged  
sites ( $\sim$ 0.05 mmol/g) on  
magnetite

E. Tombácz, E. Illés, A. Hajdú, I.Y.Tóth, R.A. Bauer, D. Nesztor, M. Szekeres, I. Zupkó, L. Vékás, PPChE, 58, 3-10, 2014.

# pH-dependent colloidal stability

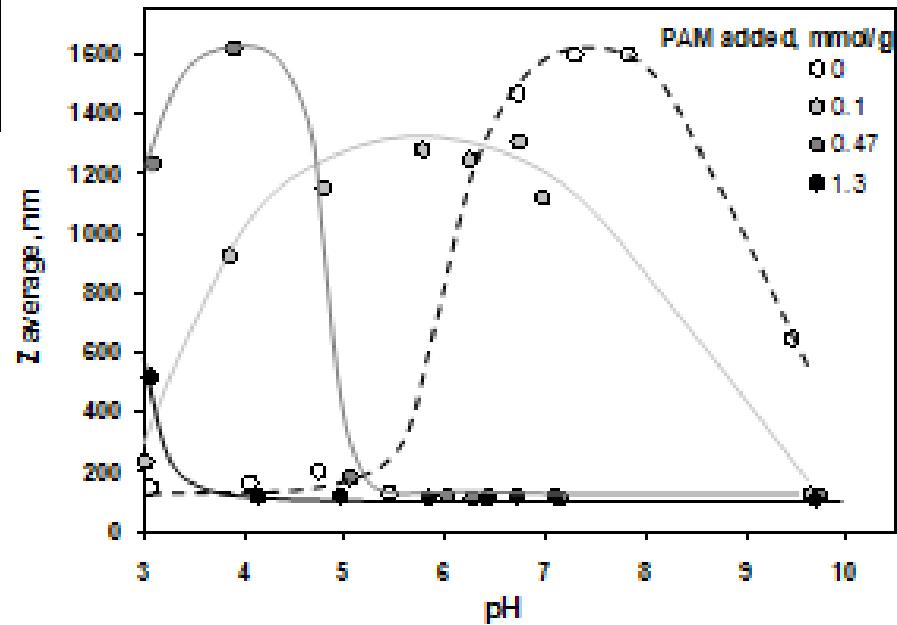


- the **isoelectric point (IEP)** shifts with increasing loading of any polycarboxylate
- low amount  $\Rightarrow$  **aggregation**
- amount high enough  $\Rightarrow$  **stabilization** over broad range of  $\text{pH} \sim 3-10$

I.Y.Tóth, E.Illés, R.A. Bauer, D.Nesztor, I.Zupkó, M.Szekeres, E.Tombácz, Langmuir, 28( 2012), 16638–16646..

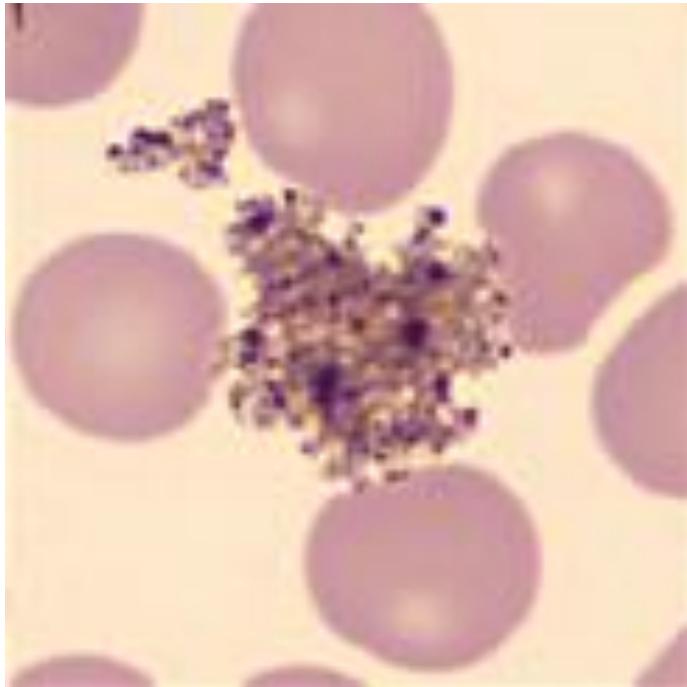
I.Y. Tóth, E..Illés, M. Szekeres, E.Tombácz, *Journal of Magnetism and Magnetic Materials*, 380:168-174 (2015)

>>> increasing pH >>>

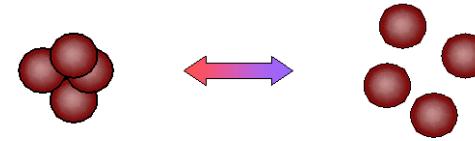


# Particle aggregation

in the blood



Smear from whole blood mixed with SPIONs



Behavior in non-uniform magnetic field



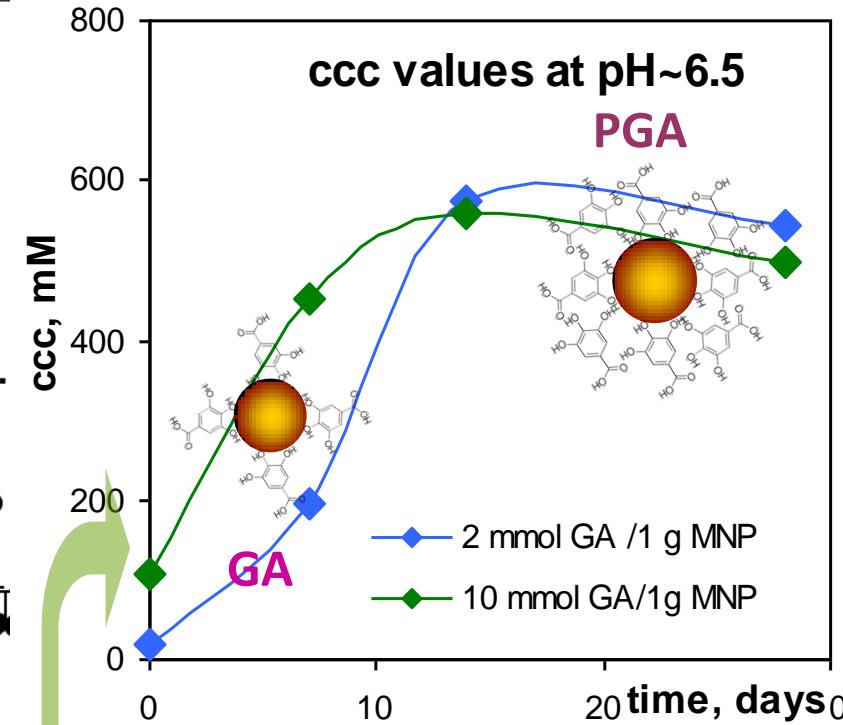
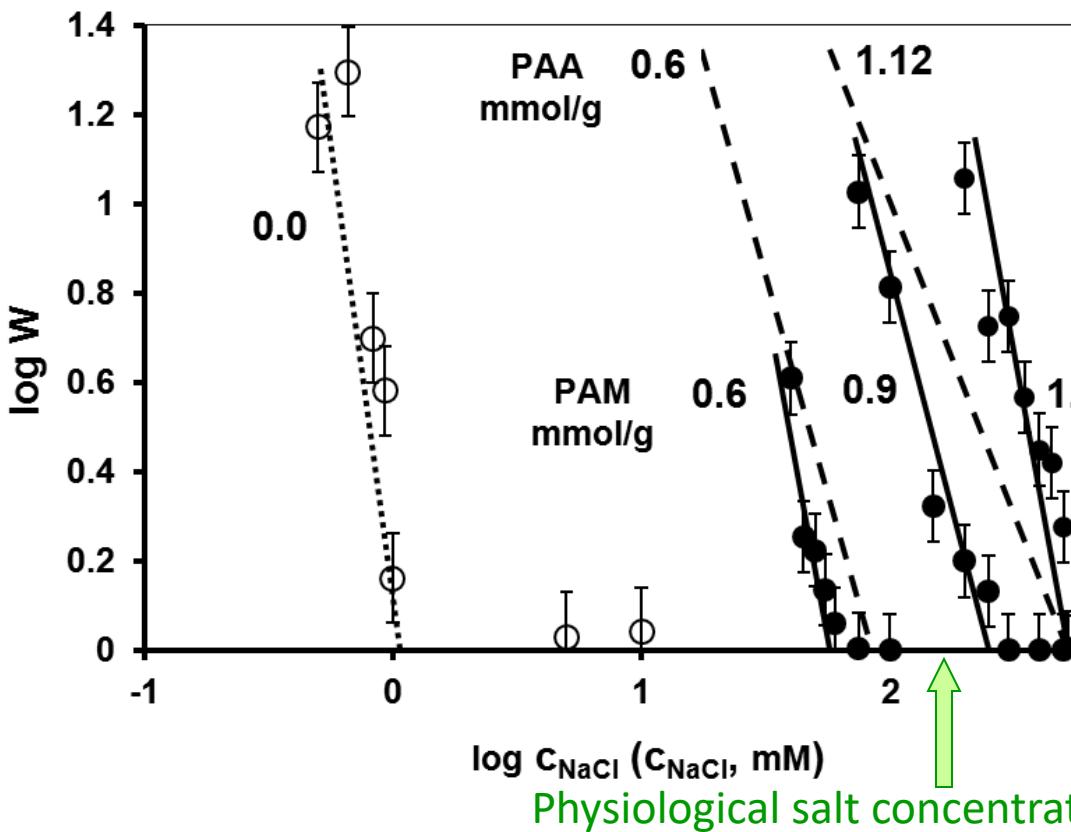
Photo of the Month November 2010  
[http://magneticmicrosphere.com/index.php?potm\\_date=11.2010](http://magneticmicrosphere.com/index.php?potm_date=11.2010)

Well coated superparamagnetic nanoparticles make beautiful ferrofluids (right), while less stable ones (left) agglomerate in high salt concentrations (e.g., blood!) under the influence of an applied magnetic field. This movie is from Prof Etelka Tombacz at the University of Szeged in Hungary (2010).

# Resistance against electrolytes

⇒ critical coagulation electrolyte concentration (ccc)

Stability plot: log w vs. log c



Enhanced stabilization due to thick, overcharged protective layers

A.Hajdú, M.Szekeres, I.Y.Tóth, R.A.Bauer, J.Mihály, I.Zupkó,

E.Tombácz, *Colloids and Surfaces B: Biointerfaces* 94 (2012) 242.

I.Y.Tóth, E.Illés, R.A. Bauer, D.Nesztor, I.Zupkó, M.Szekeres,

E.Tombácz; *Langmuir*, 28( 2012), 16638–16646..

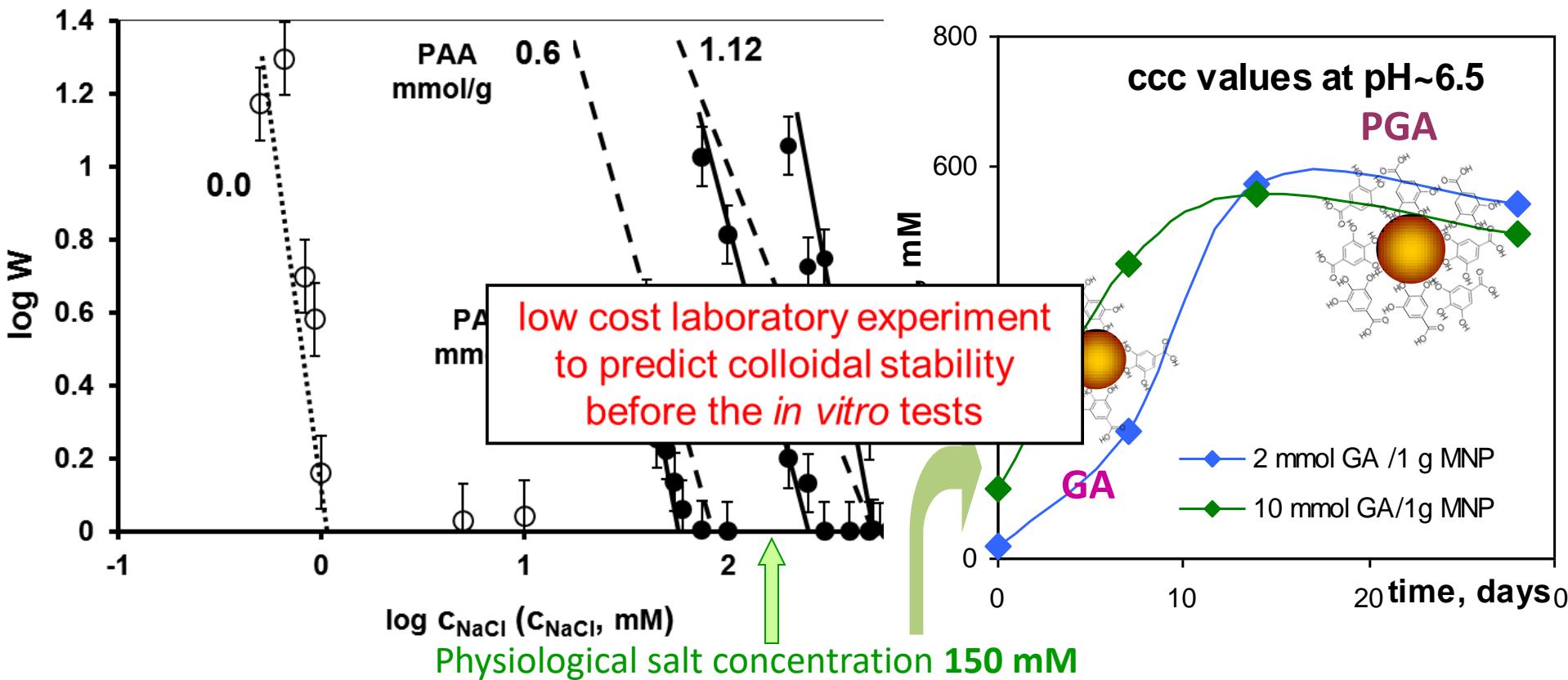
I.Y.Tóth , M.Szekeres, R.Turcu, S.Sáringér, E.Illés, D.Nesztor,

E.Tombácz, *Langmuir* 30 (2014) 15451-15461:

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I.Y.Tóth, E.Illés, R.A. Bauer, D.Nesztor, I.Zupkó, M.Szekeres,

E.Tombácz, *Langmuir*, 28(48), 16638

I.Y.Tóth , M.Szekeres, R.Turcu, S.Sáringér, E.Illés, D.Nesztor,

E.Tombácz, *Langmuir* 30 (2014) 15451-15461:

Polyacids@MNP	Added amount mmol COOH/g	Approx. CCC NaCl, mM
Naked MNP	0	1
CA@MNP	0.3*	70
PAA@MNP	1.12	500✓
PAM@MNP	1.18	500✓
PEGMA-AA@MNP	1.2	>150✓

Non-eligible for in vitro tests

## Physicochemical and colloidal characterization

Eligible for in vitro tests

CCC < ~ 500 mM CCC ≥ ~ 500 mM

6. Colloidal stability: salt tolerance (CCC)

$ \zeta  < 35 \text{ mV}$	$ \zeta  > 35 \text{ mV}$
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5. Characterization of particle charge: electrokinetic potential ( $\zeta$ )

Non-passivized	Chemically stable (Passivized)
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4. Chemical stability: dissolution/corrosion test

3. Quantitative optimization of formulation; testing dilution resistance of coating

Outer-sphere	Inner-sphere surface complex
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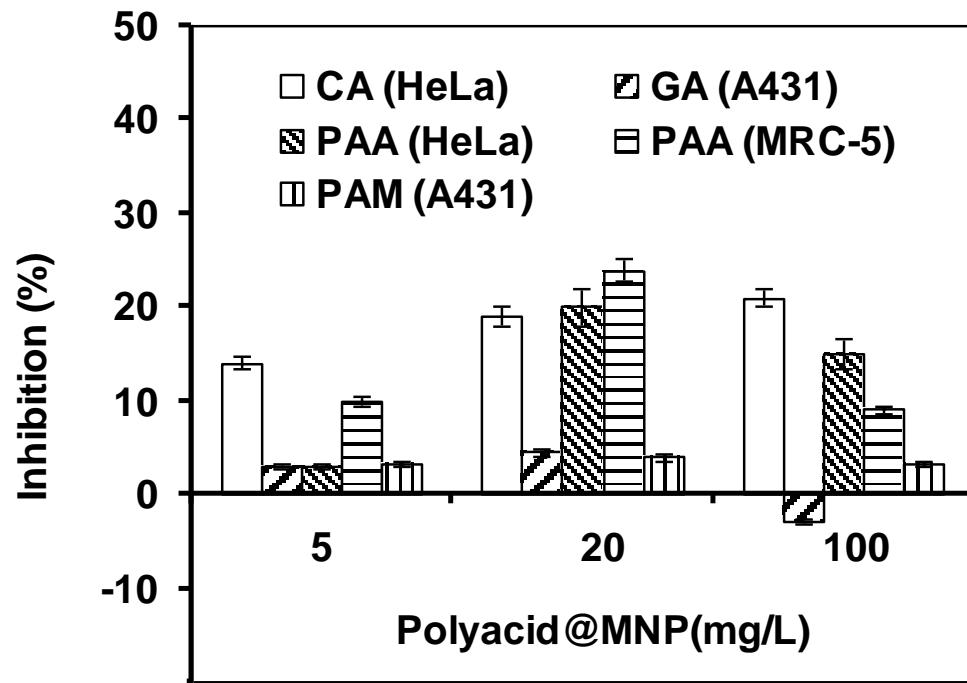
2. Strength of interaction, quality of surface bonds (surface analytics, e.g. ATR-FTIR)

Non-high affinity	High-affinity isotherm
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1. Adsorption coating, qualitative and quantitative characterization (isotherm measurement)

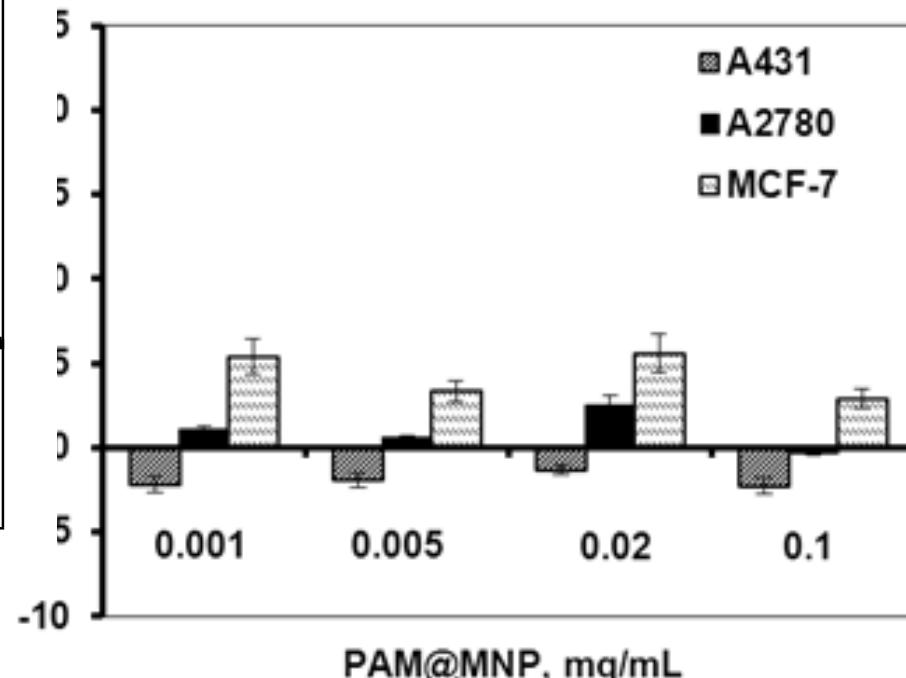
# Biocompatibility of polyacid coated MNPs

## Cell viability experiments (MTT assays)



Cell inhibition of the CA, GA, PAA and PAM-coated MNPs, added in identical concentrations to human cell cultures HeLa, MRC-5 and A431.

E. Tombácz et al. Colloidal stability of carboxylated iron oxide nanomagnets for biomedical use, *PPCE*, 2014, 58, 3-10.



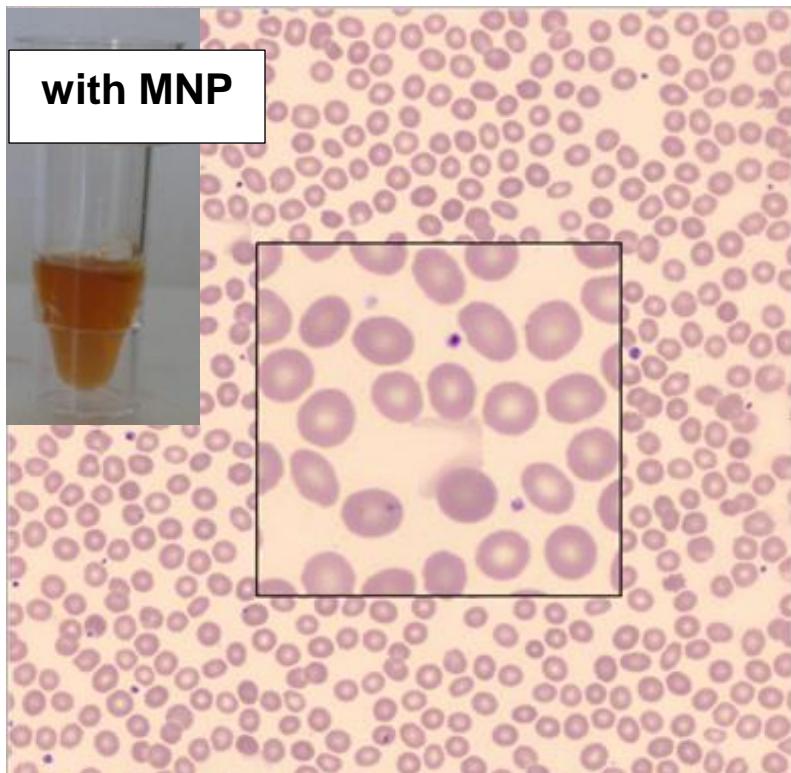
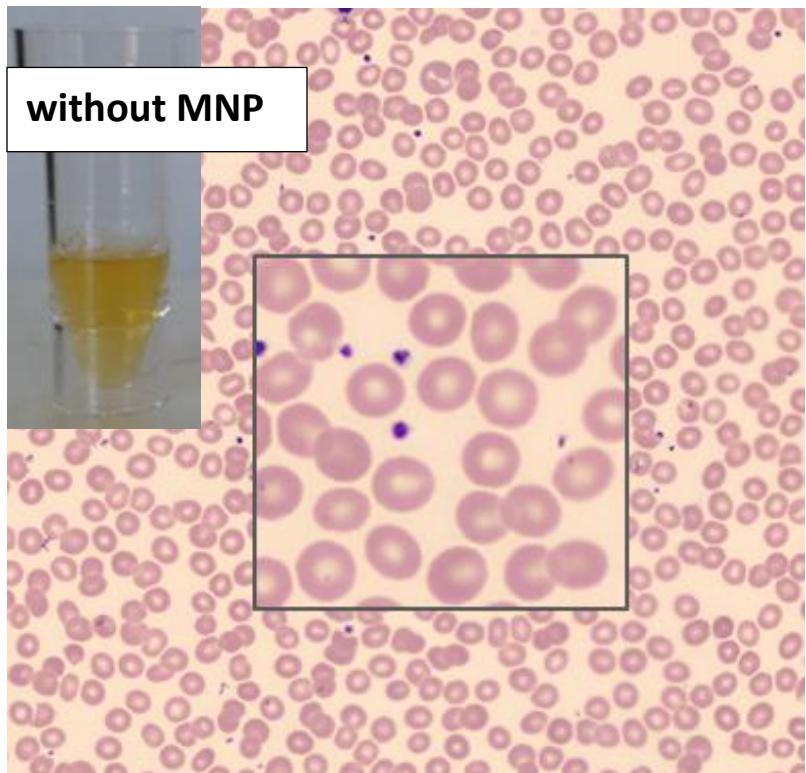
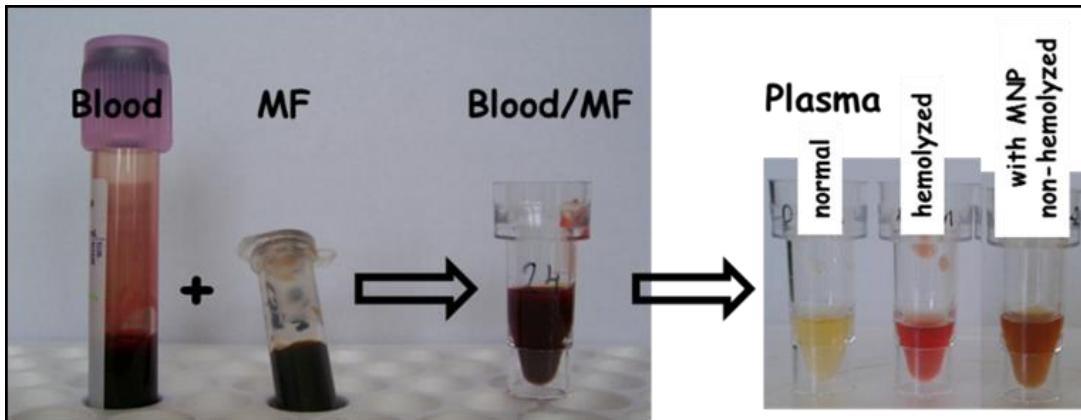
Cell inhibition of the PAM-coated MNPs, added in identical concentrations to human cell cultures MRF-7, A2780 and A431.

I.Y. Tóth, E. Illés, R. A. Bauer, D. Nesztor, M. Szekeres, I. Zupkó, E.Tombácz: Designed Polyelectrolyte Shell on Magnetite Nanocore for Dilution-Resistant Biocompatible Magnetic Fluids, *Langmuir*, 2012, 28 (48), 16638–16646.

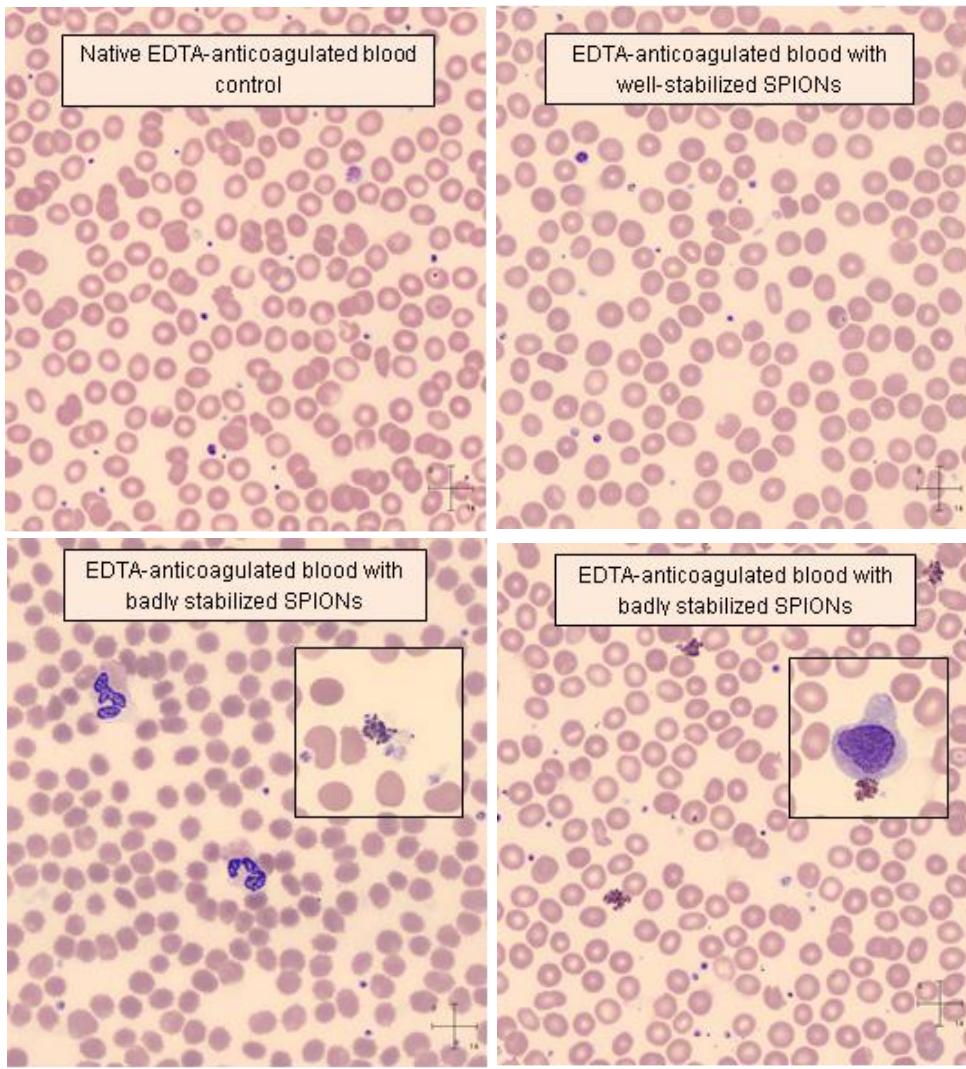
# Hemocompatibility tests for P(PEGMA-AA)@MNP

## Peripheral blood smears

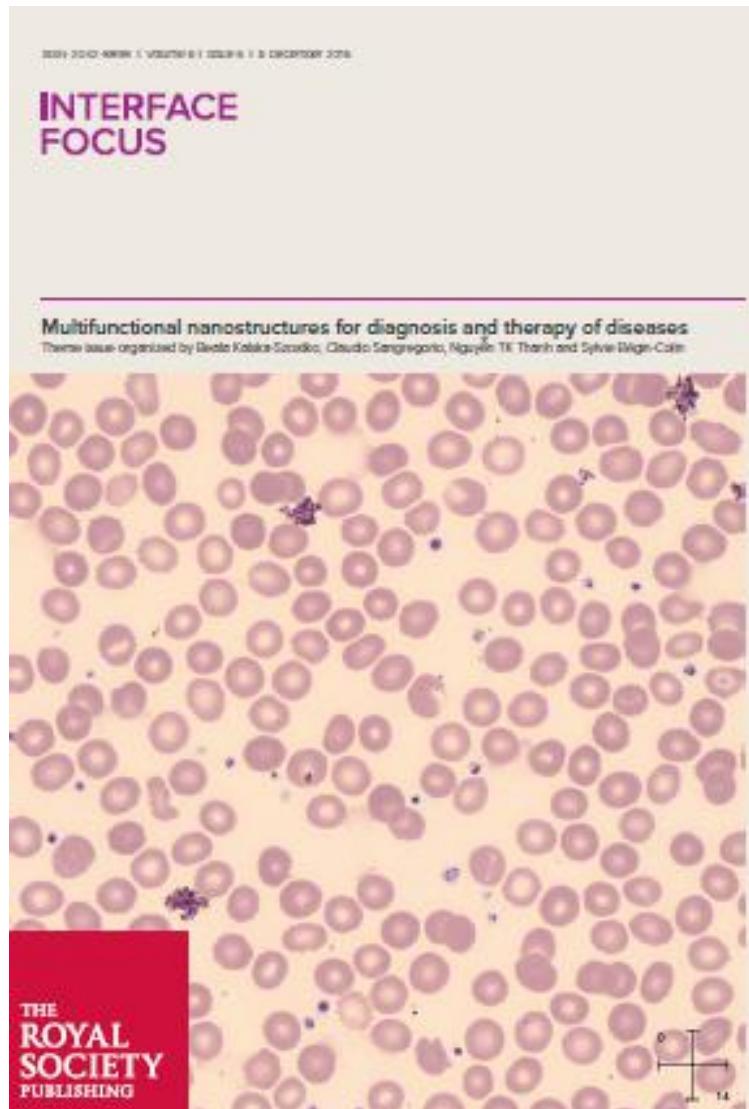
- no hemolytic activity
- no sign for thrombocyte aggregation



E. Illés, E. Tombácz, M. Szekeres, I.Y. Tóth, Á. Szabó, B. Iván: Novel carboxylated PEG-coating on magnetite nanoparticles designed for biomedical applications, *Journal of Magnetism and Magnetic Materials*, 380: 132-139 (2015)



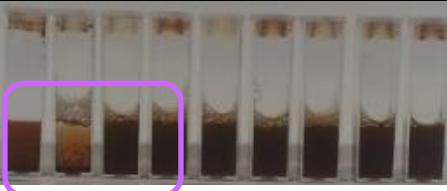
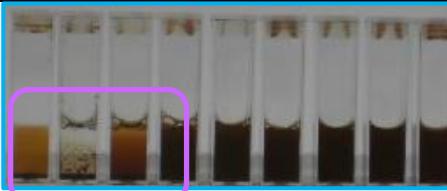
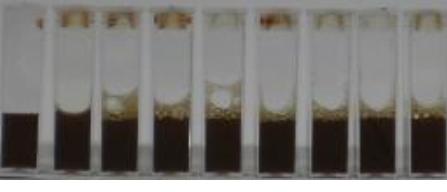
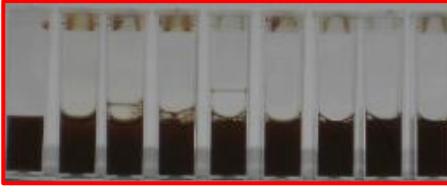
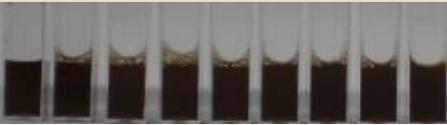
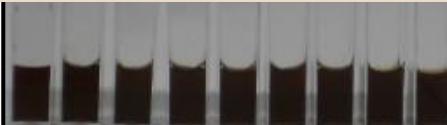
Tombácz E, Farkas K, Földesi I, Szekeres M, Illés E, Tóth IY, Nesztor D, Szabo T. (2016) Polyelectrolyte coating on superparamagnetic iron oxide nanoparticles as interface between magnetic core and biorelevant media, *Interface Focus* 6: 20160068.  
doi:10.1098/rsfs.2016.0068



### Cover image

Haemocompatibility test (blood smear) of superparamagnetic iron oxide nanoparticles (SPIONs). (Image courtesy of Katalin Farkas and Imre Földesi (Department of Laboratory Medicine, University of Szeged).)

# Interaction of carboxylated SPIONs with human plasma

	>>> human plasma in increasing concentration >>>	
mixing with	after 2 hours	after 21 hours
CA@MNP		
PAM@MNP		
The top PEG layer can not hinder interactions with proteins.		
PEGMA-AA@MNP		

Materials	Protein adsorption	Cell adhesion
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## Hydrophilic materials

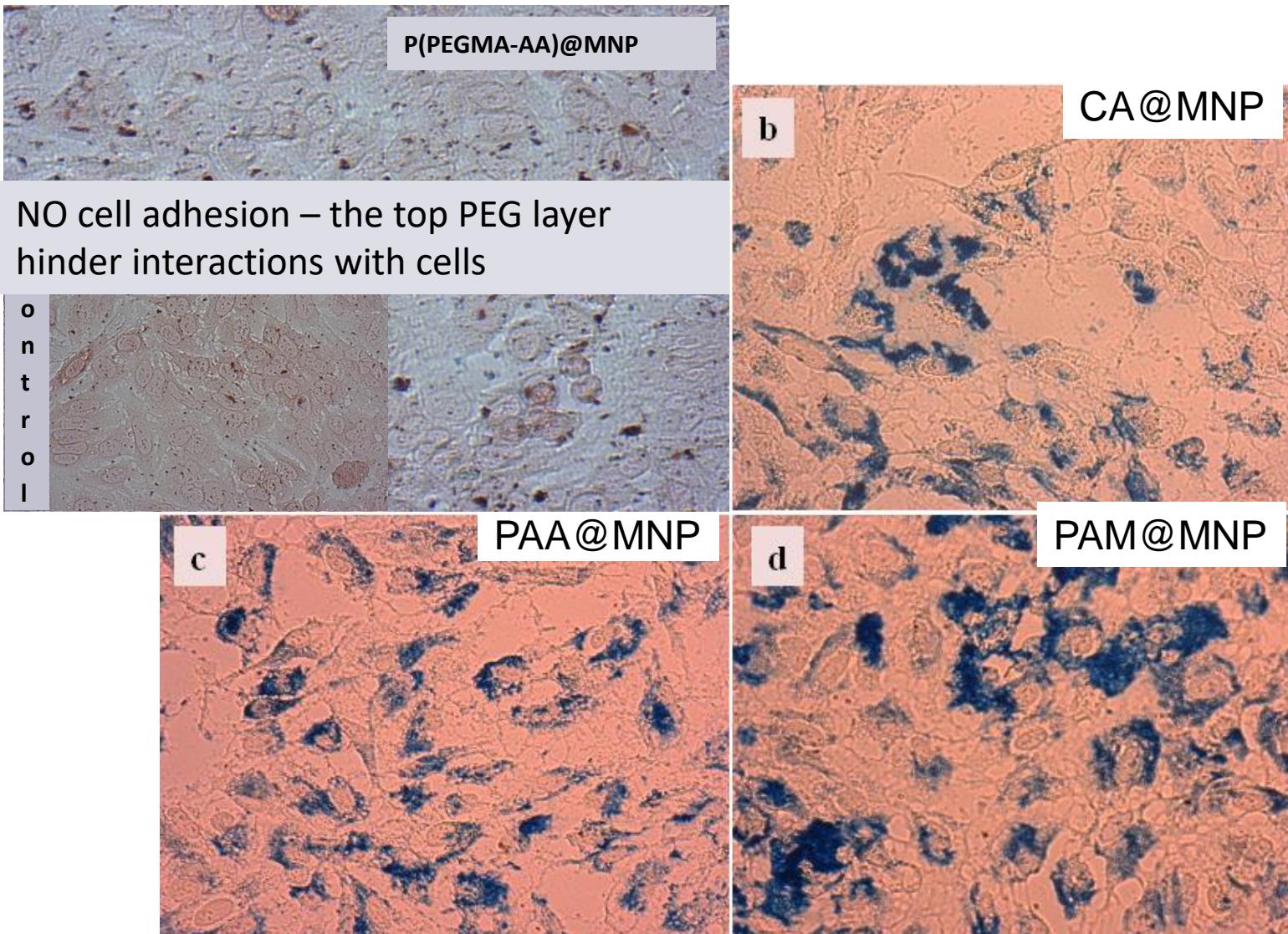
### PEG-based materials

PS-g-PEGMA and PMMA-g-PEGMA [92]	Yes	No
PEG-poly(phosphonate) terpolymer [93]	Yes	No
PLL-g-PEG [94,95]	Yes	Yes
PEGMA [96,97]	Yes	No
PPEG <sub>x</sub> Lys [98]	Yes	No
POEGMA [99–102]	Yes	Yes
PEO-PU-PEO [61,103–105]	Yes	No
PEO-PPO-PEO [106]	Yes	No
PEO [33]	Yes	No

M. Szekeres, I.Y. Tóth, R. Turcu, E.Tombácz, The effect of polycarboxylate shell of magnetite nanoparticles on protein corona formation in blood plasma, *Journal of Magnetism and Magnetic Materials* DOI: 10.1016/j.jmmm.2016.11.017

S. Chen et al. Surface hydration: Principles and applications toward low-fouling/nonfouling biomaterials, *Polymer* 51 (2010) 5283-5293.

Prussian blue staining of HeLa cells (**a**) incubated with 7.35 mg/mL concentration dispersions of CA@MNP (**b**), PAA@MNP (**c**), and PAM@MNP (**d**)



# Grafting proteins to SPIONs via carboxylate anchoring groups

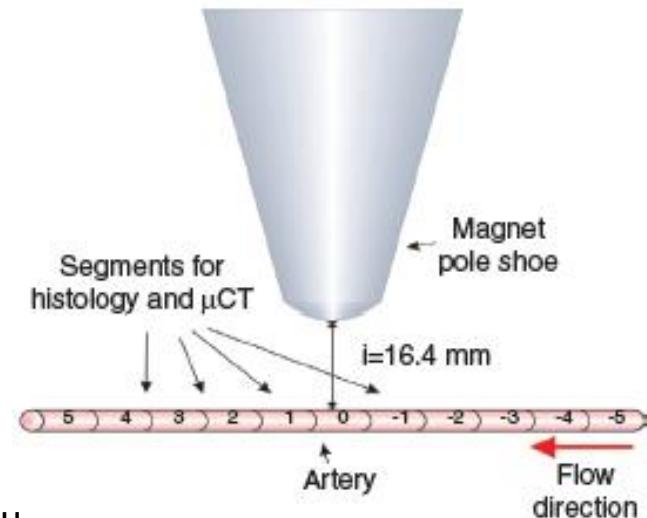
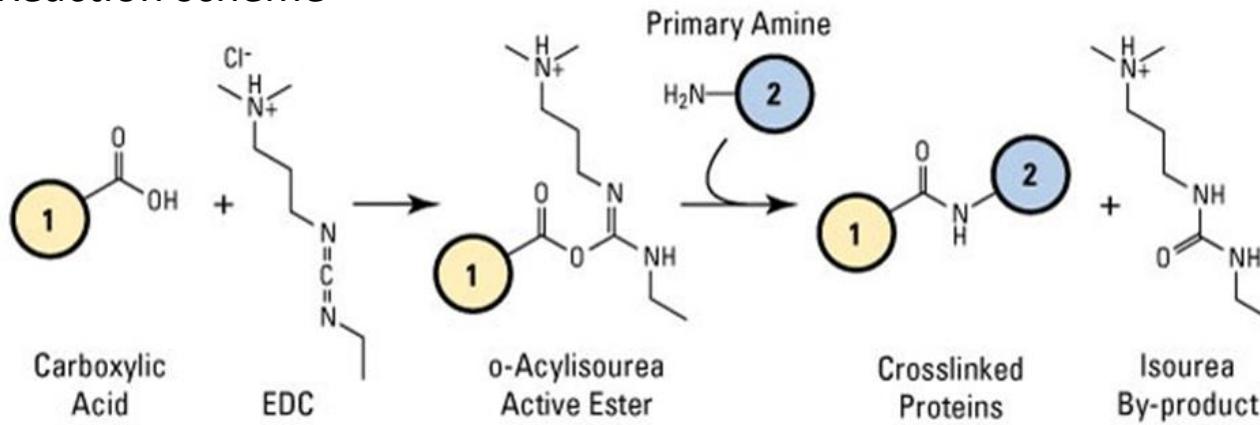
Magnetic targeting  
local application of drug

Recombinant tissue plasminogen activator (**tPA**, Actilyse®) was coupled to polyacrylic acid-co-maleic acid coated SPIONs (**PAM@MNP**) using an amino-reactive activated ester reaction (**EDC/NHS – carbodiimide/N-hydroxysuccinimide activator**).

Covalent linkage significantly **improves the reactivity and long term stability** of the conjugated **SPION-tPA system** compared to simple adsorption.

RP Friedrich, J Zaloga, E Schreiber, IY Tóth, E Tombácz, S Lyer, C Alexiou,  
*Nanoscale Research Letters* (2016) 11:297

Reaction scheme



in SEON group of **Prof. Christoph Alexiou** (University Hospital Erlangen, Section for Experimental Oncology and Nanomedicine, Erlangen, Germany)

# Potential for theranostic application

## Magnetic Resonance Imaging (MRI) – venous admin

Coated SPIONs

$r_2$  relaxivities

CA@MNP

$156 \text{ mM}^{-1}\text{s}^{-1}$

PAA@MNP

**$396 \text{ mM}^{-1}\text{s}^{-1}$**

PAM@MNP

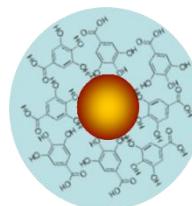
**$440 \text{ mM}^{-1}\text{s}^{-1}$**

PEGMA-AA@MNP

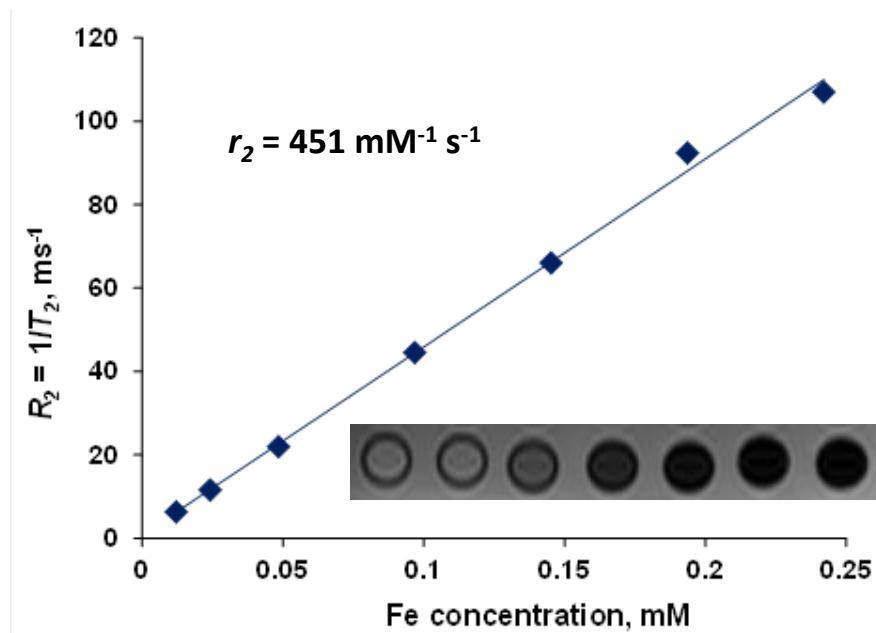
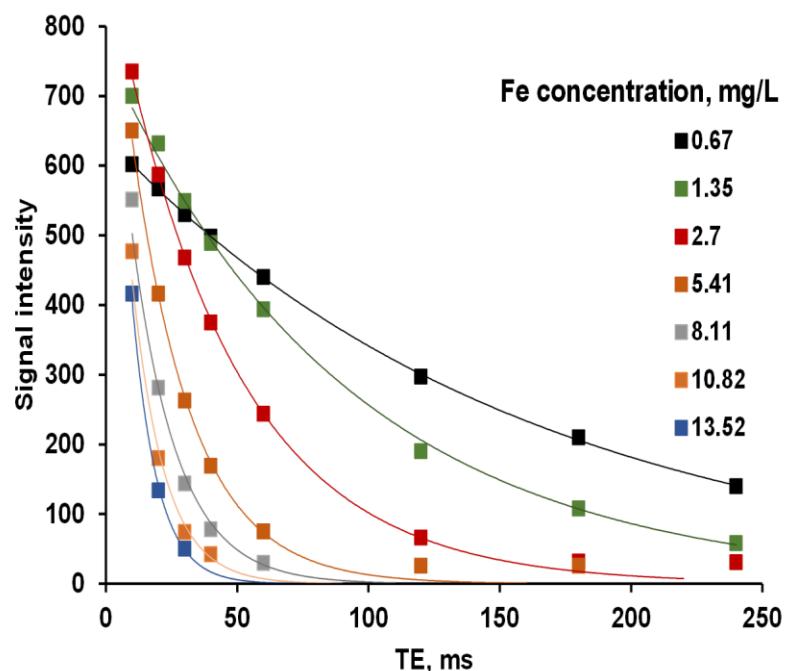
**$451 \text{ mM}^{-1}\text{s}^{-1}$**

Resovist (dextrane coated MNP)

$306 \text{ mM}^{-1}\text{s}^{-1}$



GE Excite HD (1,5 T, EUROMEDIC DIAGNOSTICS SZEGED KFT.)



Jedlovszky-Hajdú, E. Tombácz, I. Bánya, M. Babos, A. Palkó, *Journal of Magnetism and Magnetic Materials*, 2012, 324, 3173-3180;  
Szekeres M, Illés E, Janko C, Farkas K, Tóth IY, Nesztor D, Zupkó I, Földesi I, Alexiou C, Tombácz E, *J Nanomed Nanotechnol.* 2015, 6: 252.;  
E. Illés, E. Tombácz, M. Szekeres, I.Y. Tóth, Á. Szabó, B. Iván, *Journal of Magnetism and Magnetic Materials*, 380: 132-139 (2015)

# Potential for theranostic application

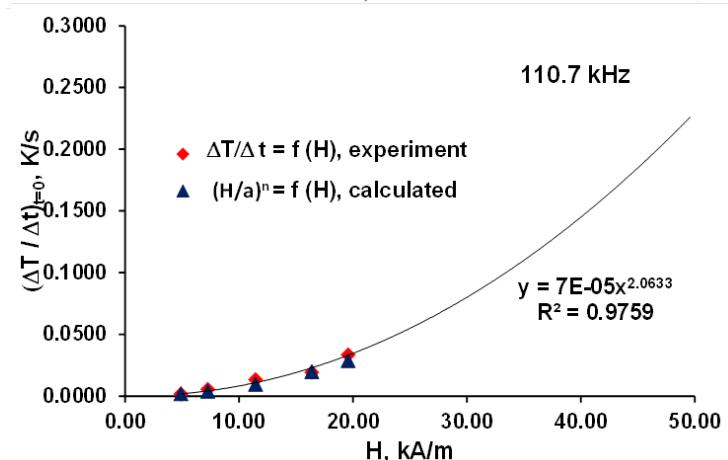
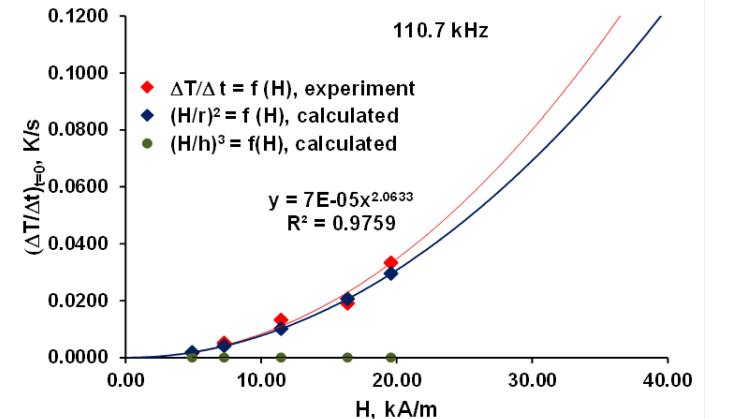
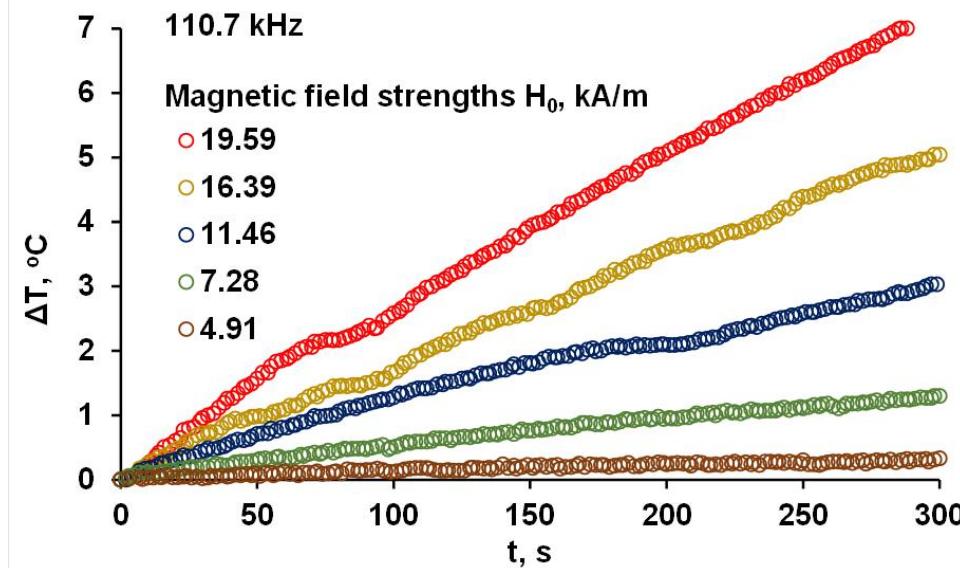
## Magnetic hyperthermia

AC magnetic field applied  
– heat release measured

## RADIOMAG (COST TD1402)

<http://www.cost-radiomag.eu/>

Heating curves (10 g/L P(PEGMA-AA)@MNP)



SAR (specific absorption rate) - no significant effect of coating on SAR

14.2 - 2.2 W/g at 110.7 kHz ( $H_0=19.6-4.9$  kA/m)

17.4 - 8.72 W/g at 329 kHz ( $H_0=13.1-5.5$  kA/m)

# Summary



~~CA~~  
PAA  
PAM  
PEGMA-co-AA



The optimization of carboxylated/PEGylated coating on SPION cores results in chemically and colloidally stable products with hydrophilic, negatively charged, non-fouling surface and anchoring sites for biofunctions.



HeLa



## MRI

r <sub>2</sub> relaxivities	mM <sup>-1</sup> s <sup>-1</sup>
PAA@MNP	396
PAM@MNP	440
PEGMA-AA@MNP	451
Resovist	306

## Teranostics - prospects

### Hyperthermia



### Magnetic targeting

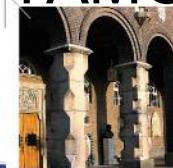
proteins  
chemotherapeutics

SAR: ~14-17 W/g  
(110 - 330 kHz)



Thanks  
my colleagues  
Márta Szekeres,  
Erzsébet Illés,  
Ildikó Tóth,  
Daniel Nesztor and  
Tamás Szabó  
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**R. Turcu** National Institute R&D for Isotopic and Molecular Technology, Cluj-Napoca, Romania  
**M. Zrinyi, A. Hajdú-Jedlovszky** Nanochemistry Group (SOTE), Budapest, Hungary  
**B. Iván** Department of Polymer Chemistry, (RCNS, HAS), Budapest, Hungary  
**I. Szalay** Institute of Physics and Mechatronics (UP) Veszprém, Hungary  
**I. Bányai, L. Novák** Department of Colloid & Environmental Chemistry (UD), Debrecen, Hungary  
**I. Földesi, K. Farkas** Department of Laboratory Medicine, (USz) Szeged, Hungary  
**I. Zupkó** Department of Pharmacodynamics and Biopharmacy, (USz) Szeged, Hungary  
**C. Alexiou** Section of Experimental Oncology and Nanomedicine, Erlangen, Germany  
for collaboration  
and grant OTKA (NK 84014) **OTKA**  
TÁMOP-4.2.2.A-11/1/KONV-2012-0047  **SZÉCHENYI TERV**  
for financial support,



and you for kind attention!

# Búcsúzik a Vizes Kolloidok Kutatócsoport

